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Perovskite Film Formation for Solar Cell Absorbers: Effects of Substrate Modification

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Introduction & Background

Pb-Based Perovskite Films as an Emerging PV Absorber

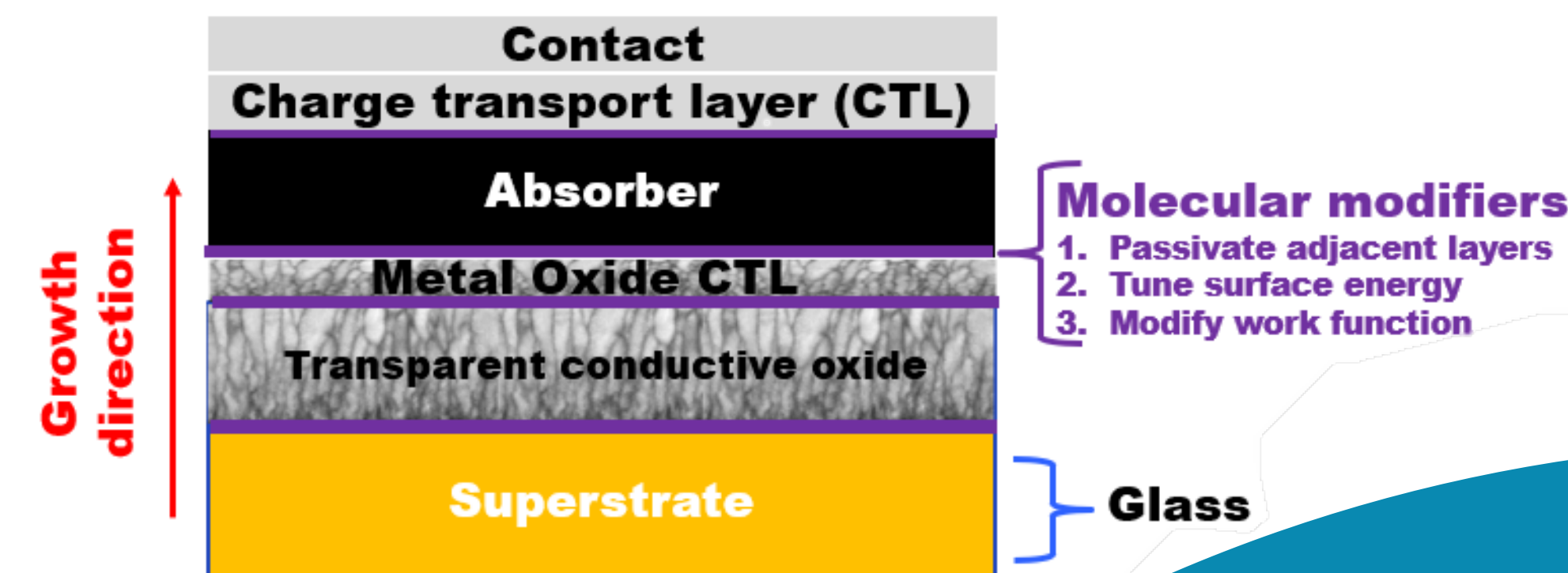
Record devices are over 20% efficient but there are materials problems that need to be solved [1].

Interfacial modification affects:

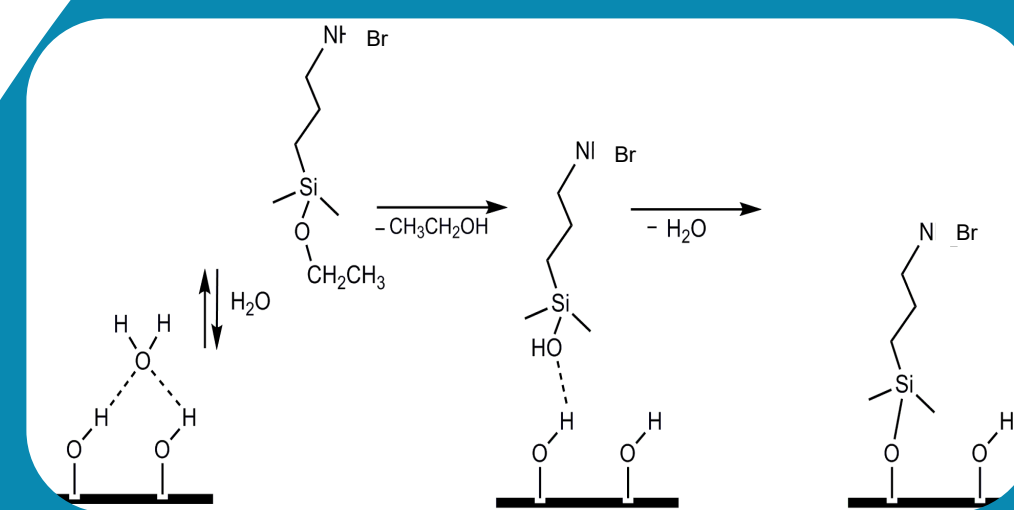
- Film uniformity
- Crystallinity
- Grain size
- Defect density

Small molecule modifiers and their positive effects on device performance. Data from references [2] and [4].

Modifier	Stack Structure	V _{oc} [V]	J _{sc} [mA cm ⁻²]	PCE [%]
Bromobenzoic Acid (Br-BA) [2]	ITO/NiO ₂ /MAPbI ₃ /PCBM/bis-C ₆₀ /Ag	1.07	19.1	15.3
	ITO/NiO ₂ /Br-BA/MAPbI ₃ /PCBM/bis-C ₆₀ /Ag	1.11	21.7	18.4
(3-Aminopropyl)triethoxysilane (APTES) [4]	FTO/SnO ₂ /MAPbI ₃ /Spiro-OMeTAD/Au	1.065	20.84	14.69
	FTO/SnO ₂ /APTES/MAPbI ₃ /Spiro-OMeTAD/Au	1.16	21.23	17.03



Goal: Improve stability of perovskite absorbers through small molecule modification of perovskite-TCO interface.



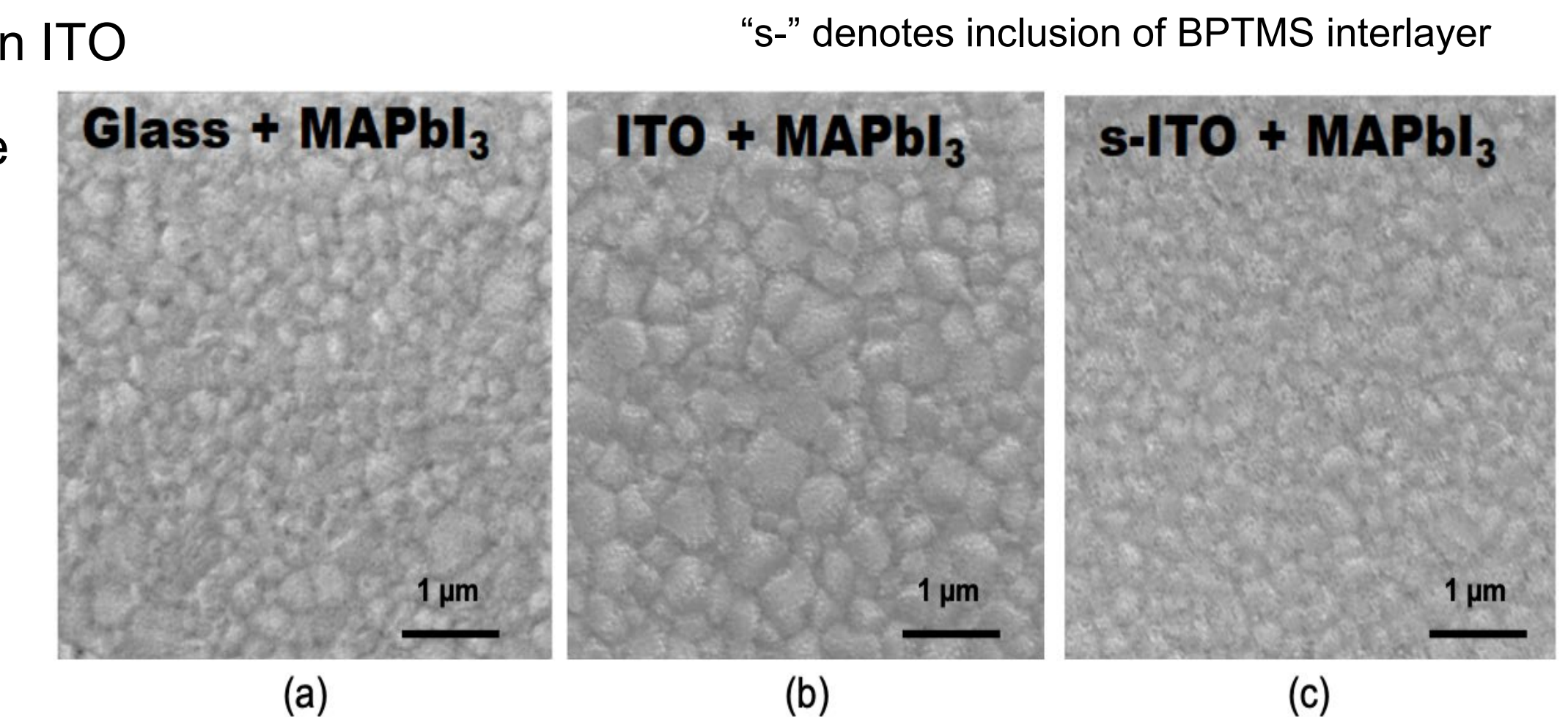
Chemical process of small molecule modification of a metal oxide.

Device properties and performance are directly tied to perovskite film properties [3].

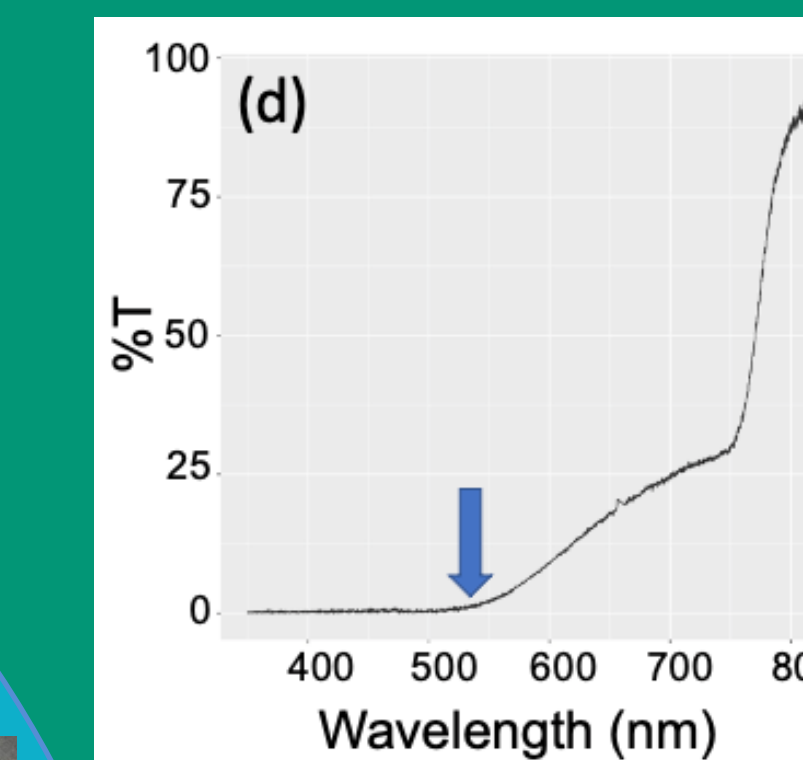
Interfacial Modification and Stability

BPTMS and MAPbI₃ Grain Growth

- BPTMS leads to growth of smaller grains on ITO
- MAPbI₃ grains on s-ITO comparable in size to those grown on glass



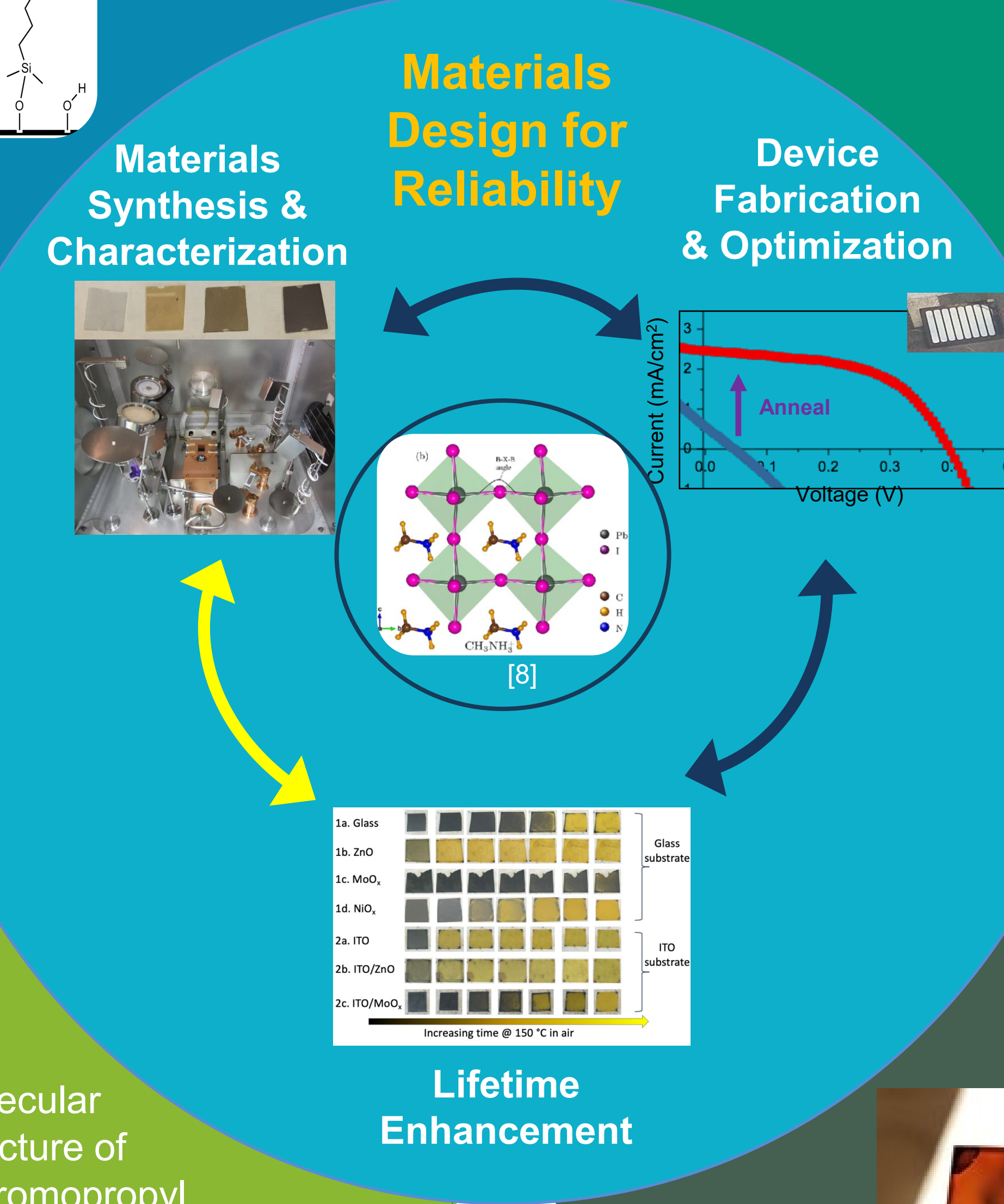
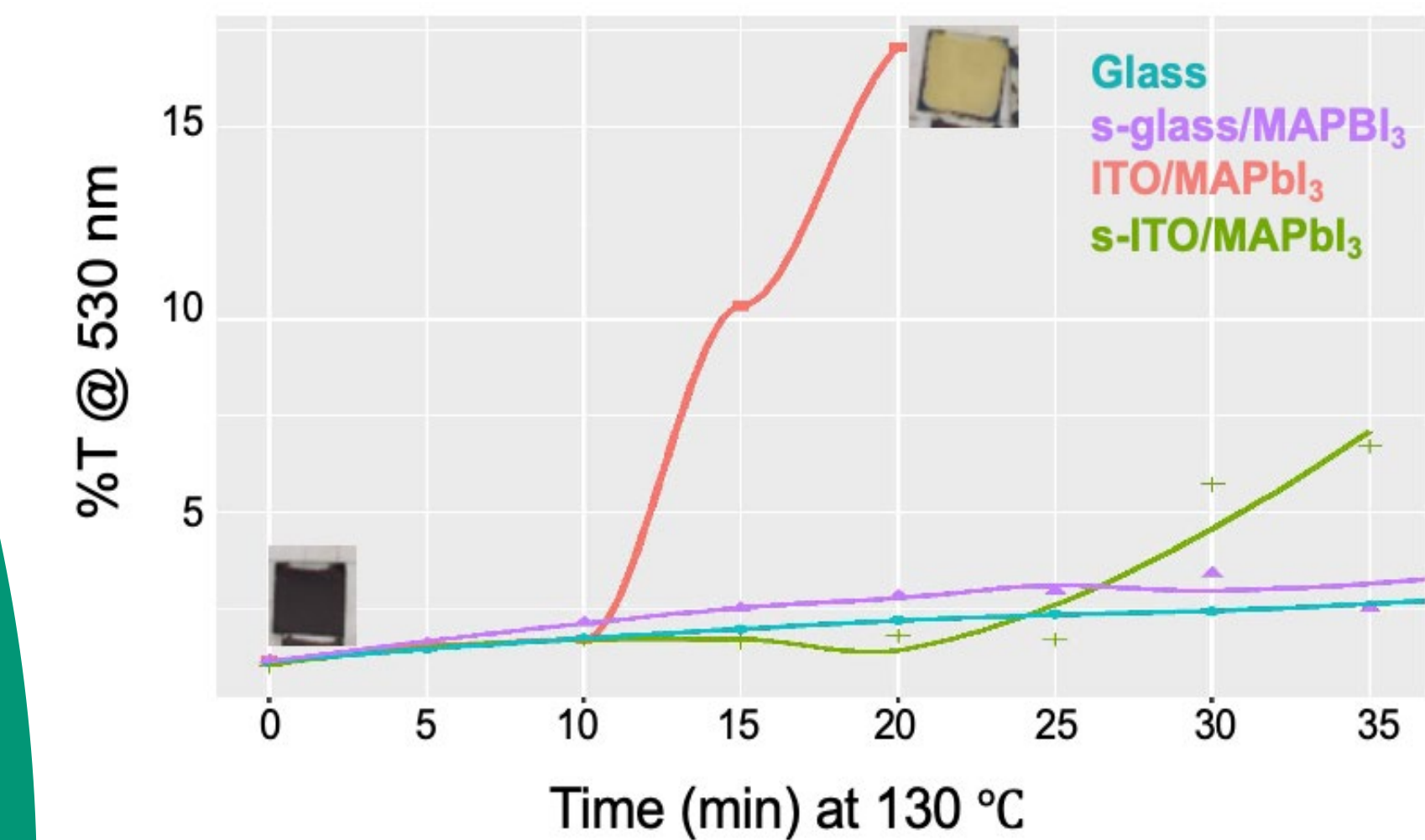
Results: BPTMS passivates the perovskite-TCO interface, affecting both the film morphology and degradation profile of the perovskite.



UV-Vis spectrum of MAPbI₃ on s-ITO.

BPTMS and MAPbI₃ Degradation

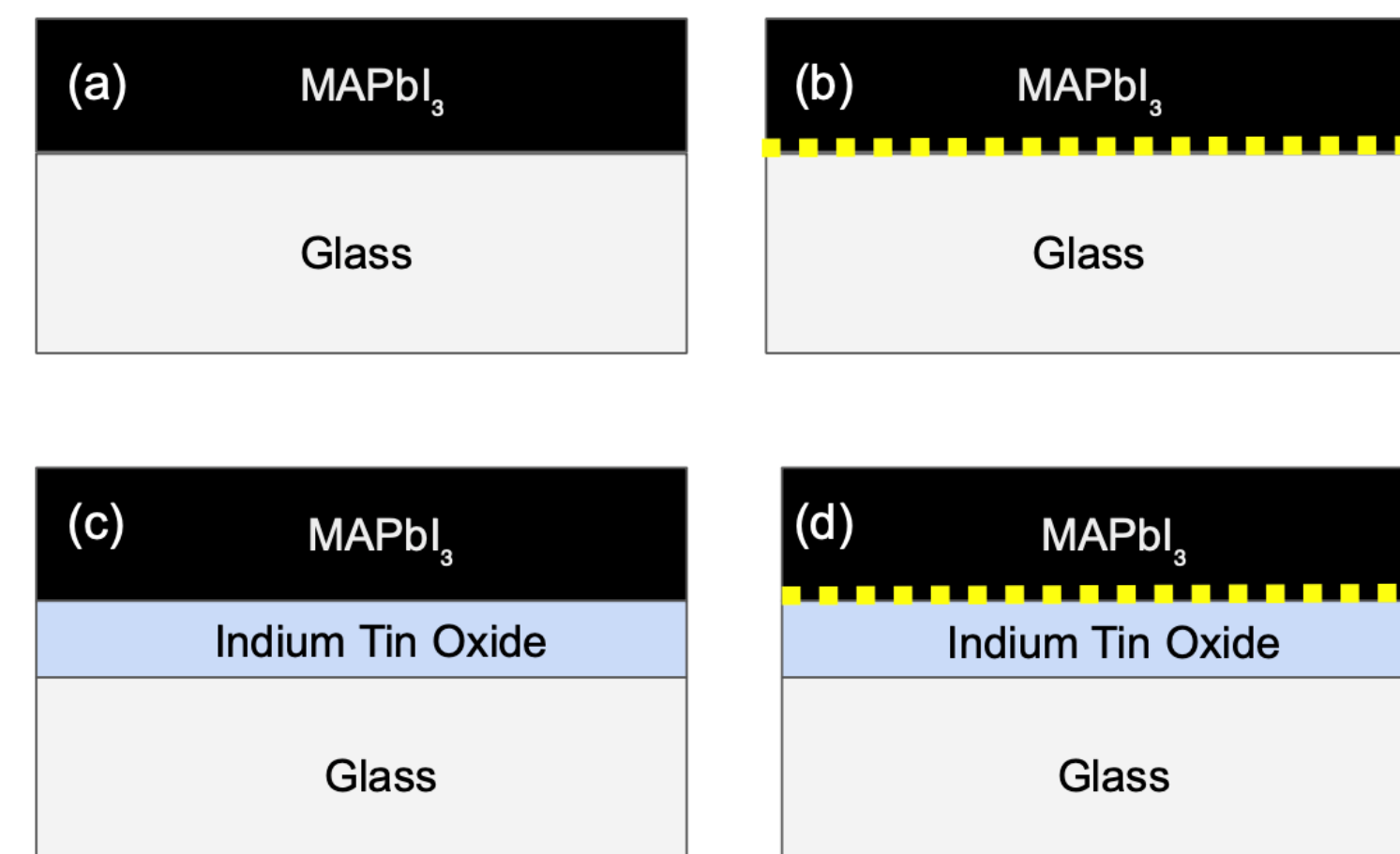
- BPTMS mitigates MAPbI₃ degradation on ITO (in green) compared to the unmodified control (in red)
- Decouples effects of grain size from interfacial chemistry in terms of stability



Experimental Flow

1. Clean substrates
2. Deposit organofunctional silane [6,7]
3. Spin coat MAPbI₃ and anneal [5]
4. Characterization and degradation

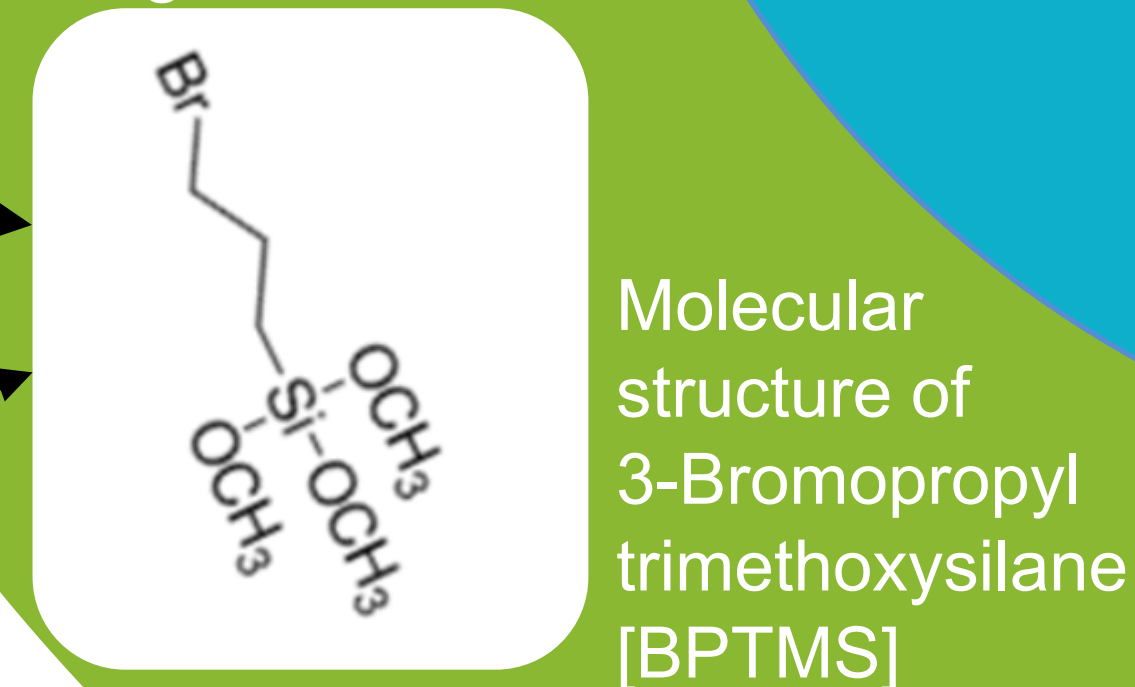
Schematic overview of film stacks



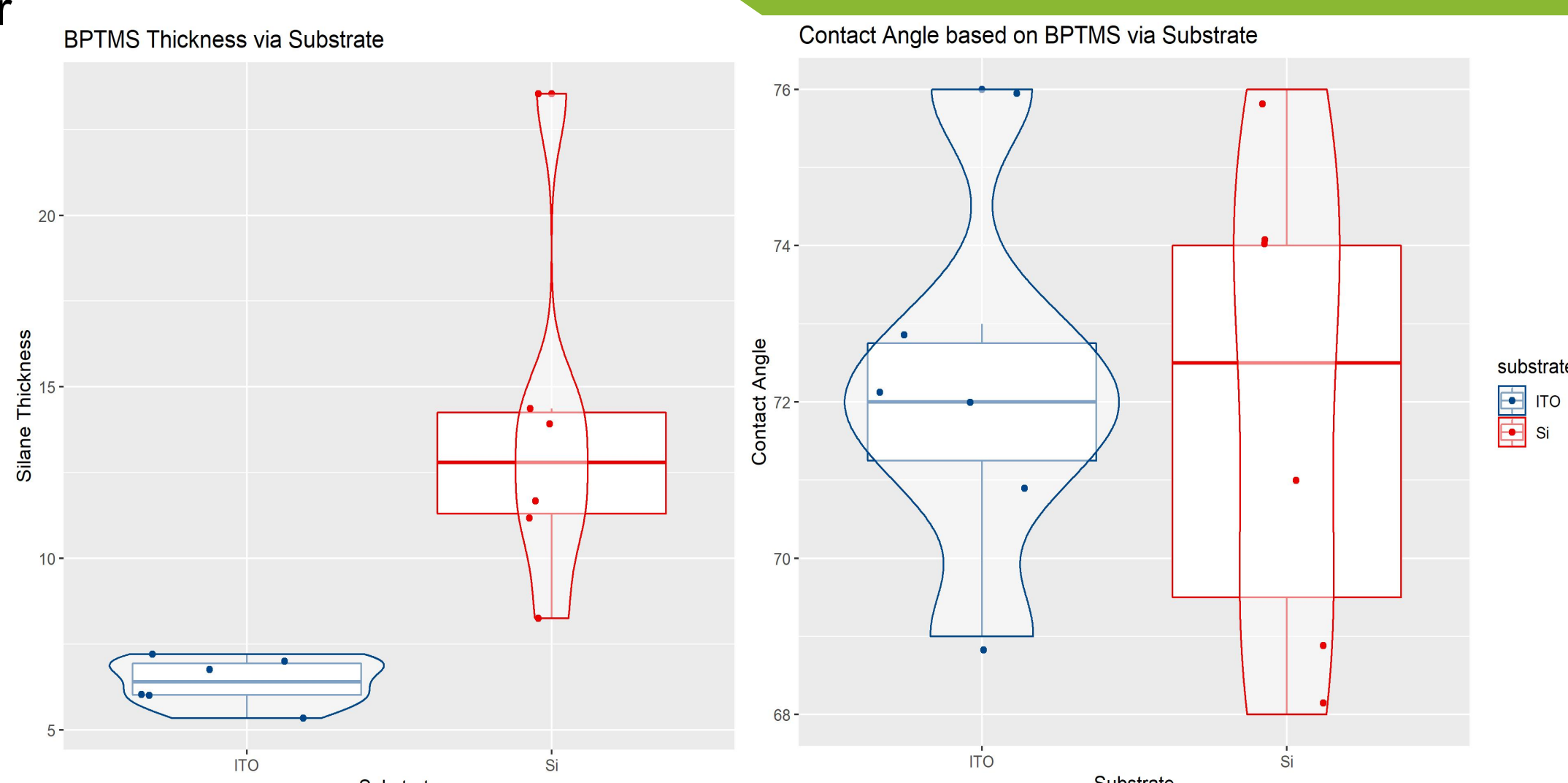
Dashed line indicates organofunctional silane layer

Substrate	Thickness, Å	Contact Angle, degrees
Silicon	11 ± 1	62 ± 3
	24 ± 1	81 ± 2
ITO	34 ± 2	84 ± 1
	6 ± 1	72 ± 2

Approach: Deposition of MAPbI₃ on bare and silane-modified substrates to systematically investigate effects of a BPTMS interlayer on perovskite degradation.



Molecular structure of 3-Bromopropyl trimethoxysilane [BPTMS]



Future Directions:

- Apply silanes to MOs (Metal Oxides), commonly used as PV charge transport layers
- Investigate systematically varied TCO/MO/silane combinations in half-stack degradation studies

- Further investigate silane-MO interface and its effects on the perovskite film with SE and XPS

Conclusions

- ❖ Results highlight importance of film studies under device-relevant conditions
- ❖ Organofunctional silanes used as molecular modifiers can passivate a TCO/perovskite interface
- ❖ Interfacial modifiers have multifaceted effects on perovskite film morphology and lifetime
- ❖ BPTMS forms mono- to multilayers on SiO₂ and a monolayer on ITO.
- ❖ As silanes bond to surface hydroxyl groups, this is likely due to differences in properties of the oxide surface, specifically the availability and/or spacing of the hydroxyl groups

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References

- 1 Nature Energy, vol. 4, pp. 1, Jan. 2019.
- 2 ChemSusChem, 10 (2017), 3794-3803. DOI: 10.1002/cssc.201701262.
- 3 J. Photon. Energy 6(2), 022001 (2016), DOI: 10.1117/1.JPE.6.022001.
- 4 Journal of Materials Chemistry A, vol. 5, no. 4, pp. 1658-1666, 2017.
- 5 ACS Appl. Energy Mater. 2020, 3, 3, 2386-2393 DOI:10.1021/acsaem.9b02052
- 6 Langmuir 29 (2013), 4057-67. DOI:10.1021/la304719y
- 7 ACS Appl. Mater. Interfaces 9 (2017), 17620-17628. DOI: 10.1021/acsaami.7b02638
- 8 AIP Advances, vol. 9, no. 8, p. 085123, 2019.